

Oxidation Charts For Iron, Manganese & Hydrogen Sulfide

Iron Oxidation

Oxidant	Amount needed to oxidize 1 ppm Fe
Oxygen (O ₂)	0.14 ppm
Chlorine (Cl ₂)	0.62 ppm
Ozone (O ₃)	0.86 ppm
Potassium Permanganate (KmnO ₄)	1.00 ppm
Iodine (I ₃)	3.60 ppm
Hydrogen Peroxide	0.50 ppm

Manganese Oxidation

Oxidant	Amount needed to oxidize 1 ppm Mn
Oxygen (O ₂)	0.21 ppm
Chlorine (Cl ₂)	2.00 ppm
Ozone (O ₃)	1.60 ppm
Potassium Permanganate (KmnO ₄)	2.70 ppm
Iodine (I ₃)	7.20 ppm
Hydrogen Peroxide	1.00 ppm

Hydrogen Sulfide Oxidation

Oxidant	Amount needed to oxidize 1 ppm H ₂ S
Chlorine (Cl ₂)	3.00 ppm
Potassium Permanganate (KmnO ₄)	6.00 ppm
Iodine (I ₃)	10.8 ppm
Hydrogen Peroxide	1.05 ppm
Ozone (O ₃)	2.00 ppm

Allow 20 – 40 minutes for complete oxidation to occur depending on temperature, pH and oxidant used.



Have questions? Call us at 888-600-5427 and speak with one of our WQA Certified Master Water Specialists. Visit us online www.CleanWaterStore.com. Email us at info@cleanwaterstore.com